



For Question No. 55 to 58



- (A) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1
- Statement 1 is True, Statement 2 is True; Statement 2 is NOT a correct explanation for Statement 1 (B)
- (C) Statement – 1 is True, Statement – 2 is False
- (D) Statement – 1 is False, Statement – 2 is True
- Statement I: Addition of bromine to trans-2-butene yields meso-2, 3-dibromo butane. 55.

Statement II: Addition of bromine to an alkene is an electrophilic addition.

(2001)

Statement I: Dimethyl sulphide is commonly used for the reduction of an ozonide of an alkene to get the 56. carbonyl compound.

Statement II: It reduces the ozonide giving water soluble dimethyl sulphoxide and excess of it (2001)evaporates.

57. What would be the major product in the following reaction?



(2001)



Consider the following reaction $H_3C - CH - CH - CH_3 + Br \longrightarrow X + HBr$



Identify the structure of the major product X:

(2002)

- H₃C CH CH CH₂ D CH₃
- $H_3C \overset{\bullet}{C} CH CH_3$
- (**D**) H₃C CH CH CH₃
- Identify a reagent from the following list which can easily distinguish between 1-butyne and 2-butyne. 59.
 - (A) bromine, CCl₄

- H2, Lindlar catalyst

- dilute H₂SO₄, HgSO₄
- (D) ammonical Cu₂Cl₂ solution
- Identify, X Y and Z in the following synthetic scheme and write their structures.

$$CH_{3}CH_{2}-C\equiv C-H \xrightarrow{\text{(i) NaNH}_{2}} X \xrightarrow{H_{2}/Pd/BaSO_{4}} Y \xrightarrow{Alkaline\ KMnO_{4}} Z$$

Is the compound Z optically active? Justify your answer.

(2002)

(2002)